lecture 20.1.2011

we had so far:

- binding in clusters and their appearances in mass spectra
 d) metallic bonding
- photoelectron spectroscopy

today

- more photoelectron spectroscopy
- optical properties of metal (jellium) clusters

the jellium model





calibration magnetic bottle spectrometer with Au₁⁻



photoelectron spectra from Ag_N⁻



comparison with the jellium model



pronounced energy gaps after 8 and 20 electrons, but there are more gaps due to lifting of degeneracy





comparison PES with jellium calculations

Density of states from KS single-particle energy eigenvalues



G. Wrigge, M. Astruc Hoffmann, and B. v. Issendorff PRA65, 063201(2002)

Al₂₀ PES vs. calc.

in this case: mainly one isomer contributes

dashed curve: measured spectrum

Akola et al., PRB 62, 13 216 (2000)



close-lying isomers



here: all three isomers contribute to the experimental spectrum

PES on coinage metal clusters







chemically similar systems may have similar PE spectra. Exception: Gold, due to relativistic effects

> Hannu Hakkinen, Michael Moseler, Oleg Kostko, Nina Morgner, Margarita Astruc Hoffmann, and Bernd v. Issendorff, PRL 93 093401(2004)

Trends in photoemission threshold energies

N-dependent trends in photoemission

Cu_N⁻ PE threshold rises with increasing cluster size





Taylor, Smalley, et al., JCP 96, 3319 (1992)

Cu_N⁻ PES

threshold energies increase with N





Taylor, Smalley, et al., JCP 96, 3319 (1992)

threshold energies: electron detachment





generally:

- IPs <u>decrease</u> with increasing N
- electron detachment en.
 or EAs <u>rise</u> with increasing N

Ionization potentials (top, by R. Whetten) and electron affinities (bottom, Meiwes-Broer group) of Aluminium clusters generally:

- the IPs decrease with increasing N
- the EAs rise with increasing N

Parametrization:

$$IP(R) = WF + \alpha \frac{e^2}{R} \quad \text{with } \alpha = 3/8 \dots \frac{1}{2}$$
$$EA(R) = WF - \beta \frac{e^2}{R} \quad \text{with } \beta = \frac{1}{2} \dots \frac{5}{8}$$

 $\alpha = \beta = \frac{1}{2}$ corresponds to the charging energy of a jellium sphere, deviations arise from QM exchange and correlation



DFT calculations Fennel et al., Rostock, measurements von Issendorff et al., Freiburg





Meiwes-Broer, Appl. Phys. A55 (1992) 430; also Bergmann/Schaeffer

let us evaluate IP and EA simultaneously

$$IP(R) = WF + \alpha \frac{e^2}{R} \quad \text{with } \alpha = 3/8 \dots \frac{1}{2}$$
$$EA(R) = WF - \beta \frac{e^2}{R} \quad \text{with } \beta = \frac{1}{2} \dots \frac{5}{8}$$

 α - β is small

solve for WF:
$$WF_{cal} = \frac{1}{2}(IP + EA) + \frac{1}{2}\frac{e^2}{R}(b-a)$$

 $WF_{cal} \approx \frac{1}{2}(IP + EA)$

thus the work function should be the mean value of IP and EA!

differences between measured IP and EA and bulk values



Small or no differences hint at free electron (or: ideal metal droplet) behaviour Meiwes-Broer in Advances in Metal and Semiconductor Clusters, Vol. 1 M. Duncan, Ed., JAI Press Inc., 1993

optical properties through confined electrons



... schwingen gemeinsam im Laserlicht wie ein angetriebenes Pendel





photofragment spectroscopy on mass-selected silver clusters



Tiggesbäumker et al., Phys. Rev. A 48, R1749 (1993); Chem. Phys. Lett. 260, 428 (1996)

"cold plasmon"

only in metals with delocalized electrons

also see: Kreibig and Volmer, Springer



similar: Giant resonances in spherical nuclei

- electrons in metal clusters and the nuclear particles are fermions. Both oscillate in the confining potential

- therefore the same model can be used in order to describe the optical properties, e.g. the random phase approximation (RPA)
- closed-shell nuclei also show only a single absorption
- in the nuclei however the excitation energies are much higher, i.e., in the MeV range

Fig. 1: Showing total photoneutron cross sections for different isotopes of tin, together with theoretical fits based on a dynamical collective model. (After reference 5)

Lycurgus cup (400 AD)



Optical properties of small particles

color due to a small amount of gold and silver particles

King Lycurgus is dragged into the underworld by Ambrosia

"Labors of the Months" (Norwich, England, ca. 1480).



The ruby color is probably due to embedded gold nanoparticles



Fullerenes and quantum dots

-Such absorption features can also be found in other small particles - It is therefore not restricted to metal particles but occur also in particles like fullerenes or semiconductor quantum dots - It turns out that we only need to know the optical properties of the material in order to calculate the optical properties

CdSe quantum dots

excitation of the electron cloud in a small particle



- Since the ion cores are much heavier than the electrons the positive charges stay more or less at there position when a light field interacts with the particle

- The electric field of the light moves the electron cloud out of the positive background -Below the resonance the electrons move with the direction of the electric field
- At resonance the movement of the cloud has a p/2 phase difference to the electric field.
- This is called the plasmon resonance
- Above resonance the phase is 180 degrees



Gustav Mie 1868-1957

born in Rostock

Gustav Mie, Beiträge zur Optik trüber Medien, speziell kolloidaler Metallösungen Ann. Phys. 25, 377 (1908)

absorption and scattering of light by small particles

C.F. Bohren and D.R. Huffman Wiley, NY 1983

$$\omega_{Mie} = \left(\frac{1}{3} \frac{e^2 n_o}{\varepsilon_o m_e}\right)^{1/2}$$

plasmon frequency of a small particle

-Everything gets simplified when the size of the particle is much smaller than the wavelength of the light

- The relevant value is the size R compared to the wavelength λ of the light, i.e. R/ λ

 $\omega_{Mie} = \left(\frac{e^2 N_e}{\alpha \cdot m_e}\right)^{1/2}$ static polarizability



Trend with small alkalies

For decreasing cluster size the plasmon energy shifts to lower values. This is called the red-shift.

Reason: influence of the spil-out

$$\omega_{Mie} = \sqrt{\frac{n_{bg}e^2}{3m\varepsilon_o}} = \frac{\omega_p}{\sqrt{3}}$$

Mie – Plasmon

or: increasing polarizability

increasing size

polarizabilities of alkali clusters



Knight et al. Phys. Rev. B 31, 2539 (1985)

not all clusters are spherical ellipsoïdal shell model

Problems with the spherical jellium model :

- fine structure of mass spectra (even-odd alternation)
- diamagnetism of even-electron clusters with formally open jellium shells (breaking of Hund's rule)

Unlike an atom, the positively charged background can deform !



not all clusters are spherical



Quantum chemical calculations





The ab-initio cluster structures are in first order well-reproduced by the ellipsoidal shell model.
One can calculate the plasmon profiles more easily on the basis of this model.



role of deformation

- -In cluster ions the number of electrons is N-1
- Closed-shell cluster therefore appear at M_{N+1}^{+}
- Ag_{9}^{+} and Ag_{21}^{+} are closed-shell clusters and only a single absorption line appear in the optical spectra.
- In Ag₁₁⁺ the prolate deformation of the particle results in a splitting of the absorption into two components.
- According to the ellipsoidal shell model the lower energy peak should have half the oscillator strength.

Optical response of mass-selected free metallic clusters





constant-density surfaces

for electron clusters and nuclei. The density of the surface is 0.00125 atomic units (or scaled units for nuclei), which corresponds to 38% of the bulk density



Fig. 99 Beispiele für Riesenresonanzen: a) In der Anregungsfunktion für (γ,n) -Prozesse an ¹⁷⁵Lu. Für deformierte Kerne ist der doppelte Höcker charakteristisch, da in der Deformationsachse eine andere Schwingungsfrequenz auftritt als senkrecht dazu [nach Ber 75a]; b) im Spektrum von unelastisch an Aluminium gestreuten α -Teilchen [Kis 76]

Silver clusters: s- and d-electrons



complex situation in Ag clusters



technical application as fluorescent labels



the following slides on clusters in strong laser fields have not been presented in the lecture. This subject is an important topic in the Sonderforschungsbereich SFB 652.

in strong laser pulses: clusters act as strong nano antennas

Gigawatt laser pulse





Pulse length: 100 femto seconds power densities up to 10.000 Gigawatts per cm²





→ minuature plasmas

clusters in intense laser fields: nanometer-sized plasmas far from equilibrium



simulated Coulomb explosion Na₅₅



Charge distribution of Pb ions from Coulomb explosion



Coulomb explosion Pulse width dependence



mechanism for the high absorption

plasmon-enhanced ionization



dual-pulse excitation of Ag_N@He-droplet



inner vs. outer ionization



after: Ch. Siedschlag and J M. Rost PRL 93 043402 (2004)

computer experiment: laser excitation of Xe clusters





Xe₁₄₁₅



on plasmon resonance: dynamic electron acceleration



field strenght for N = 500:

140 GeV/m

Surface-Plasmon Assisted Resonant electron acceleration in Clusters SPARC

Nano electron accelerator

attosecond electron bursts

numerical experiment



Fennel et al., PRL 98, 143401 (2007)

density-temperature plane

